## X-Ray Structure of Alborixin, a New Antibiotic Ionophore

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Summary The structure of a new antibiotic, alborixin, has been determined by X-ray and chemical methods.

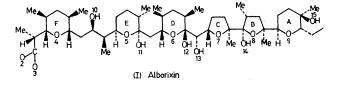
FROM cultures of a strain of *Streptomyces albus*, we have isolated a biologically active compound we named alborixin. It shows activity against gram-positive bacteria and antifungal properties which will be described elsewhere.

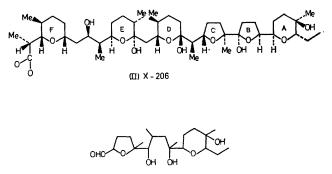
Alborixin (I) is a monocarboxylic acid,  $pK_{a}$  10.02 (MeOH). Its potassium salt,  $C_{48}H_{83}O_{14}K$ ,¶ is a crystalline solid, m.p. 209—210 °C;  $\nu_{max}$  (KBr) 3700—3100 (OH) and 1560 cm<sup>-1</sup> (CO<sub>2</sub><sup>-</sup>); m/e 922 ( $M^+$ ). The free acid (I),  $C_{48}H_{84}O_{14}$ , is an amorphous solid, m.p. 100—115 °C;  $[\alpha]_{578}^{39} - 7^{\circ}$  (c 4, acetone). Treatment of (I) by conventional methods afforded the methyl ester  $C_{49}H_{86}O_{14}$ , m.p. 67—68 °C, a triacetate,  $C_{54}H_{90}O_{17}$ , m.p. 70—75 °C, and a tetrasilylated compound. The presence of 6 OH groups (alcohols and hemiacetals) was confirmed by the mass spectra of these derivatives. A derivative corresponding to the reduction of 3 hemiacetal rings was obtained by treatment of (I) with NaBH<sub>4</sub>. It was oxidised by KIO<sub>4</sub> to give a product with m.p. 55—56 °C, to which we assign structure (III).

The structure of alborixin potassium salt was established by X-ray diffraction analysis on crystals obtained from aqueous EtOH.

Crystal data:  $C_{48}H_{83}O_{14}$ -K<sup>+</sup>, M = 922; monoclinic, a = 12.202 (4), b = 16.087 (5), c = 13.471 (5) Å,  $\beta = 102.43^{\circ}$ ,

 $D_{\rm m} = 1.5$ ;  $D_{\rm c} = 1.55$  g cm<sup>-3</sup>; Z = 2; space group  $P2_1$ . 5334 independent reflections were collected on a Siemens





(III)

computer-controlled automatic diffractometer, with Nifiltered Cu- $K_{\alpha}$  radiation. 4414 non-zero reflections were

¶ Satisfactory elementary analyses have been obtained for all compounds whose molecular formulae are given.

used in the Fourier synthesis and least-squares refinement. The potassium was found from a Patterson synthesis. However, the well known pseudo-mirror problem related to the  $P2_1$  space group, the heavy atom, and the high symmetry of the molecule led to the failure of the multisolution direct method generally used for such structures. A method, based on enantiomorph discrimination by the quartets,<sup>1</sup> and a modified tangent formula for phase refinement was then developed.\*\* The structure was refined by least squares to an R value of 0.067.

Alborixin is very similar to X-206 (II). [The addition of 4 methyl groups (on rings B, D, and F) and the exchange of an Me by an OH group do not modify the conformation and the absolute configuration of the backbone]. The backbone 'describes a path similar to that of the seam of a tennis ball,' but the hydrogen bonding and the cation co-ordination are slightly different in alborixin and X-206.

Three intramolecular hydrogen bonds stabilize the conformation:  $O(2) \cdots O(15)$ , 2.64;  $O(3) \cdots O(12)$ , 2.58; and  $O(11) \cdots O(14)$  2.78 Å. The potassium is co-ordinated to 8 oxygen atoms in a distorted cubic arrangement. The distances are:  $K \cdots O(2)$ , 2.89;  $K \cdots O(7)$ , 3.07;  $K \cdots$ O(8), 2.81; K · · · O(9), 2.76; K · · · O(10), 2.71; K · · · O(11), 2.98; K · · · O(12), 2.69; K · · · O(15), 2.76 Å.

Alborixin is a new member of the family of polycyclic polyether monocarboxylic acid antibiotics which now includes monensin, nigericin, X-537A (lasalocid), grisorixin, dianemycin, X-206, A-204A,<sup>3</sup> salinomycin,<sup>4</sup> A-23187,<sup>5</sup> septamycin,<sup>6</sup> and lysocellin.<sup>7</sup>

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\*\* Details of the method and its application will be published elsewhere.

- <sup>1</sup> J. F. Blount and J. W. Westley, Chem. Comm., 1971, 927.
  <sup>3</sup> N. D. Jones, M. O. Chaney, J. W. Chamberlin, R. L. Hamill, and S. Chen, J. Amer. Chem. Soc., 1973, 95, 3399 and references therein.
  <sup>4</sup> H. Kinashi, N. Otake, H. Yonehare, J. Sato, and Y. Saito, Tetrahedron Letters, 1973, 4955.
  <sup>5</sup> M. O. Chaney, P. V. Demarco, N. D. Jones, and J. L.Occolowitz, J. Amer. Chem. Soc., 1974, 96, 1932.
  <sup>6</sup> T. J. Petcher and H. P. Wezer, J. C.S. Chem. Comm., 1974, 697.
  <sup>7</sup> N. Otake, M. Vernurk, H. Kinschi, S. Sato, and Y. Saito, L.O.S. Chem. Comm., 1974, 697.

- <sup>7</sup> N. Otake, M. Koenuma, H. Kinashi, S. Sato, and Y. Saito, J.C.S. Chem. Comm., 1975, 92.

<sup>&</sup>lt;sup>1</sup> H. Hauptman, Acta Cryst. (A), 1974, 30, 472.